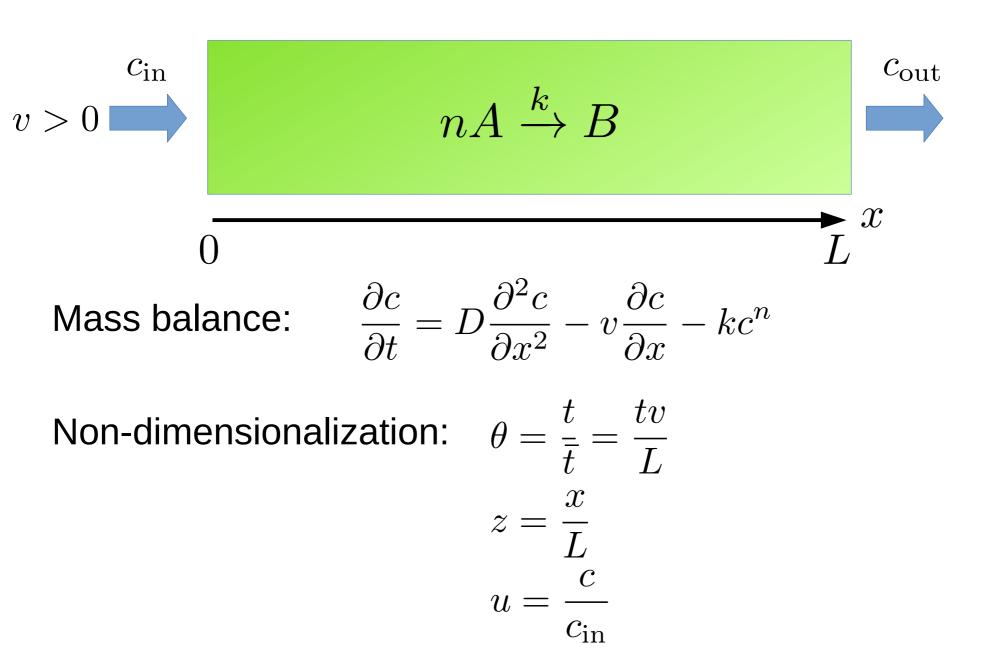
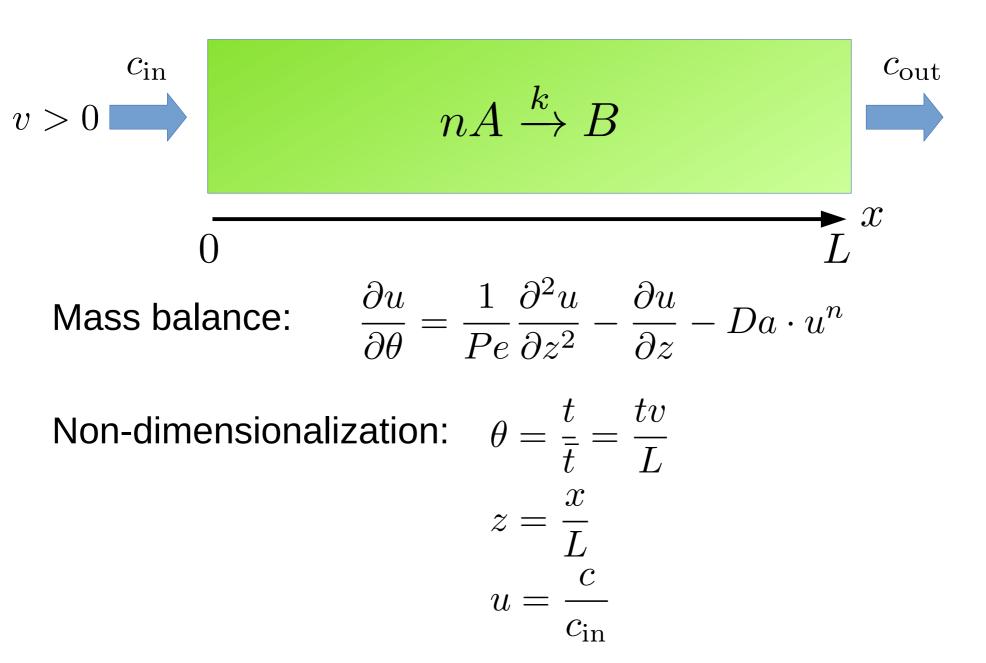
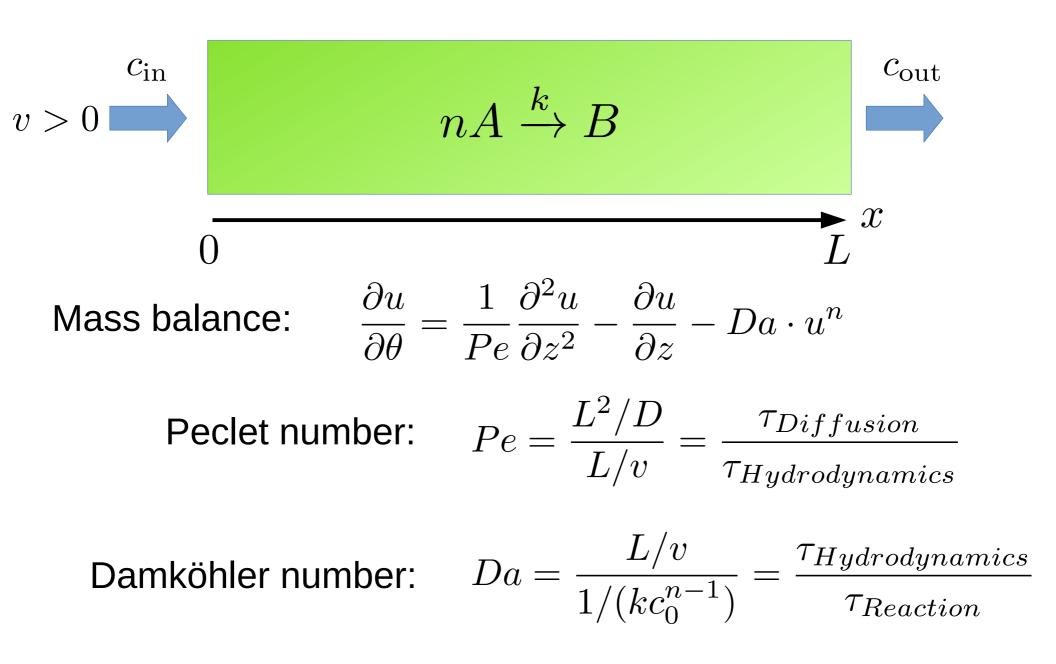
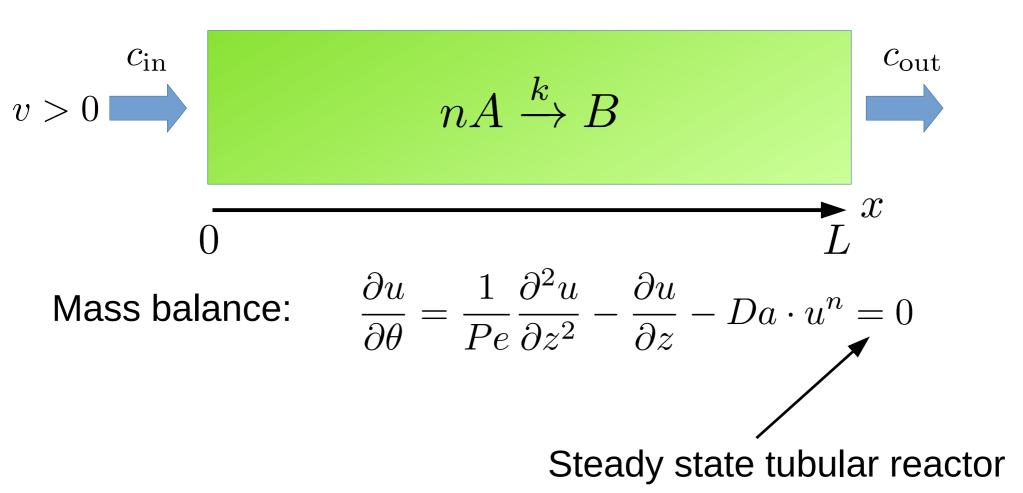


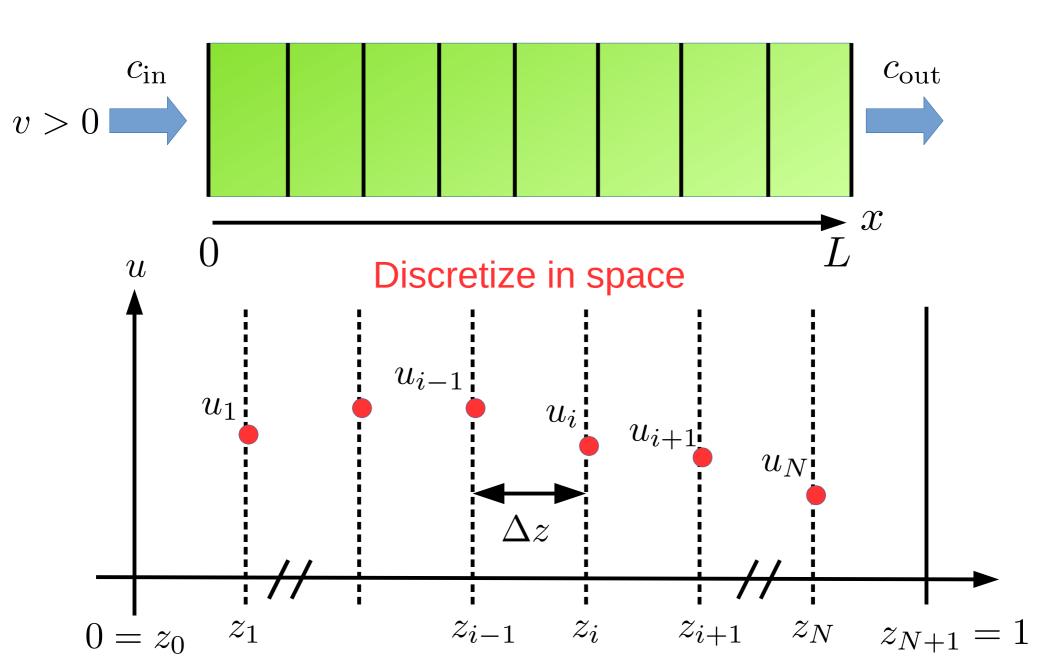
(Danckwerts)

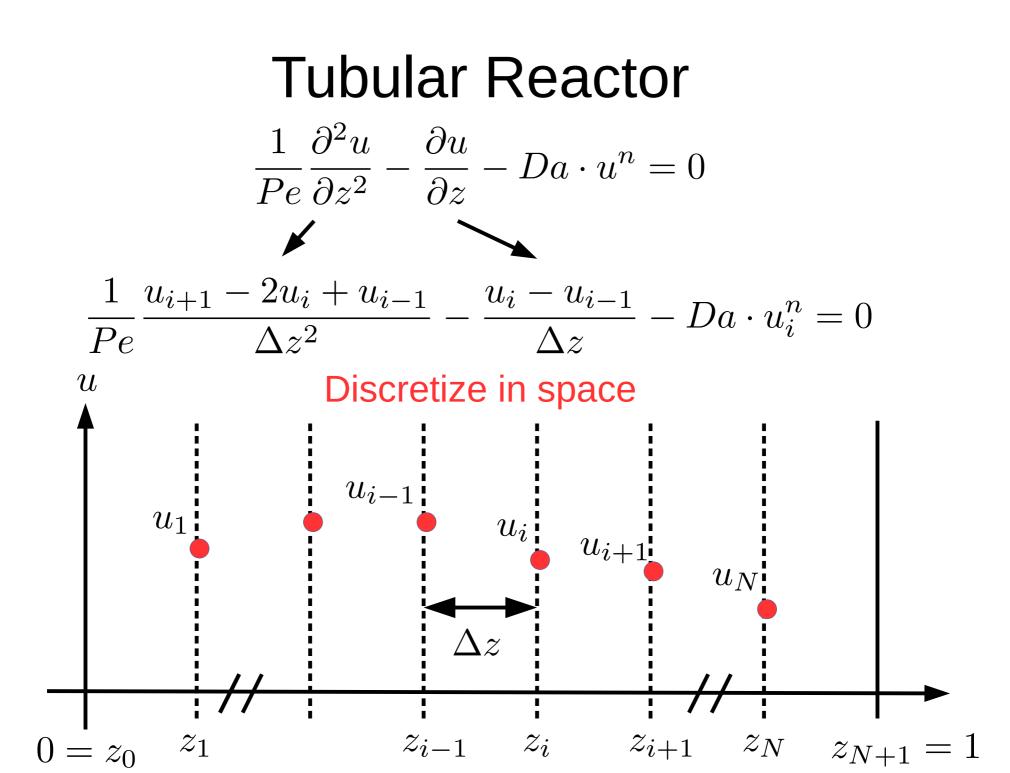


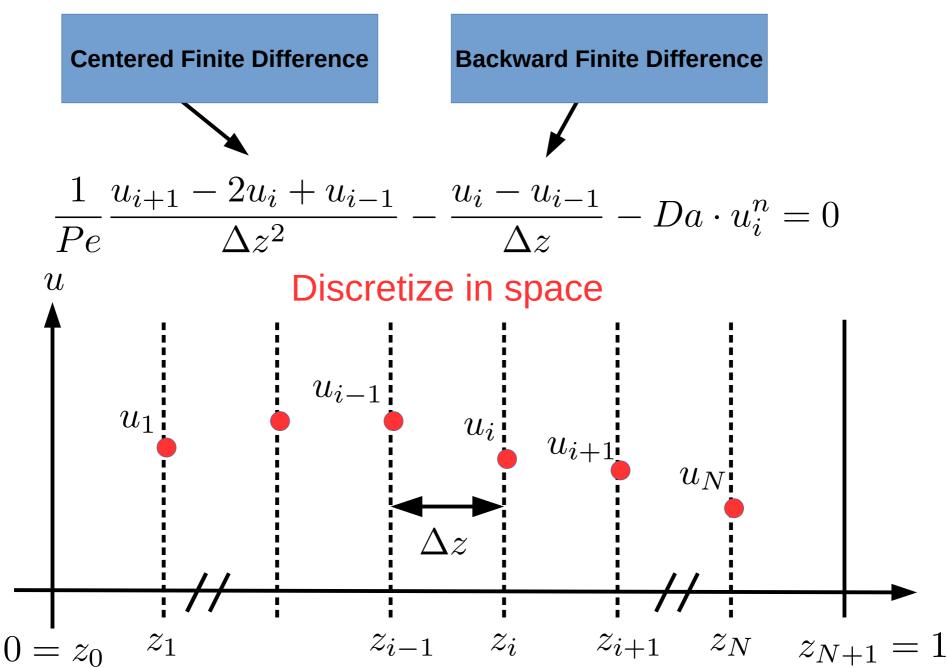


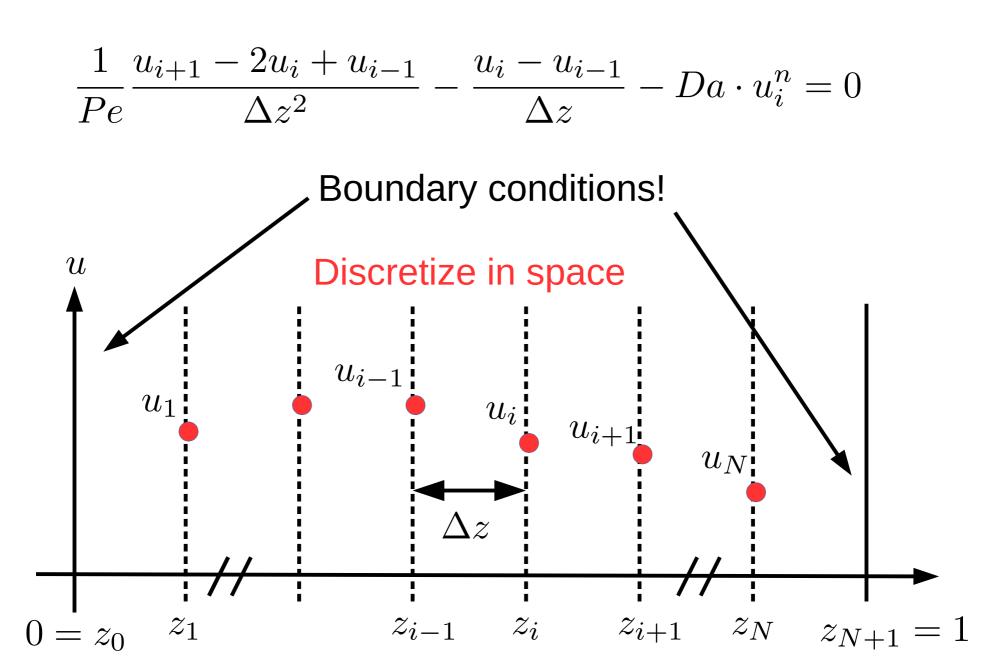


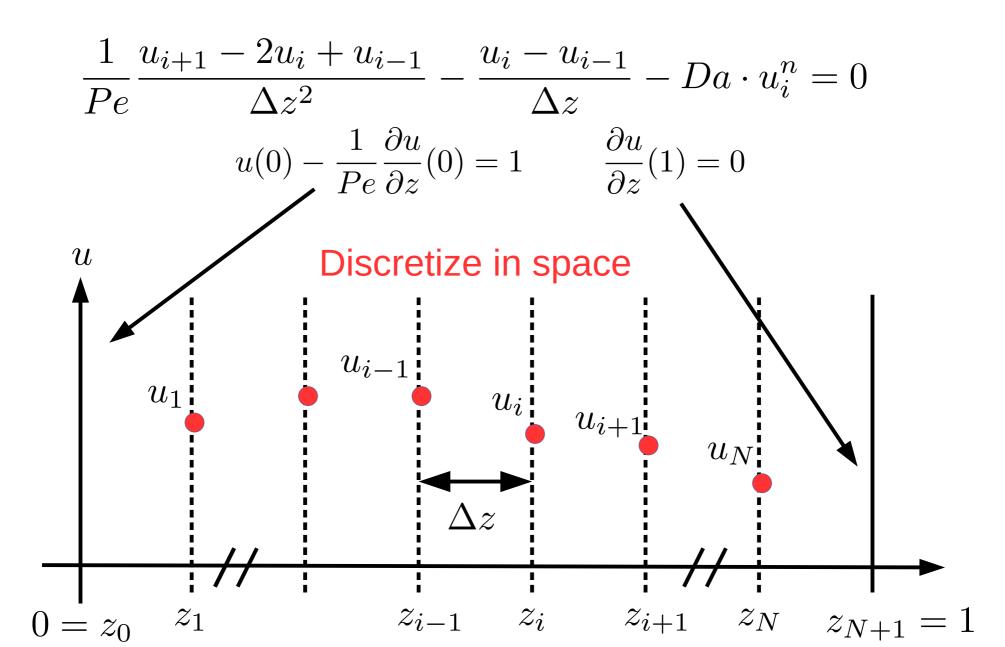


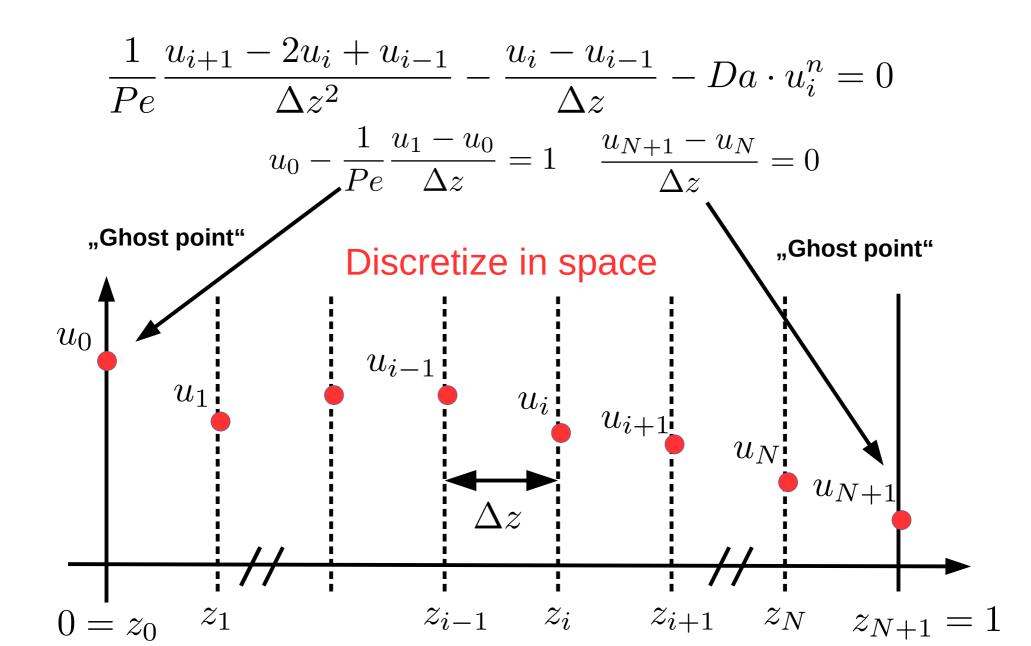












$$\frac{1}{Pe} \frac{u_{i+1} - 2u_i + u_{i-1}}{\Delta z^2} - \frac{u_i - u_{i-1}}{\Delta z} - Da \cdot u_i^n = 0$$
$$u_0 - \frac{1}{Pe} \frac{u_1 - u_0}{\Delta z} = 1 \longrightarrow u_0 = \frac{1}{1 + \frac{1}{Pe\Delta z}} \left(\frac{1}{Pe\Delta z}u_1 + 1\right)$$
$$\frac{u_{N+1} - u_N}{\Delta z} = 0 \longrightarrow u_{N+1} = u_N$$

i = 1, 2, ..., N

System of nonlinear equations!!!

Numerical Methods for Chemical Engineers

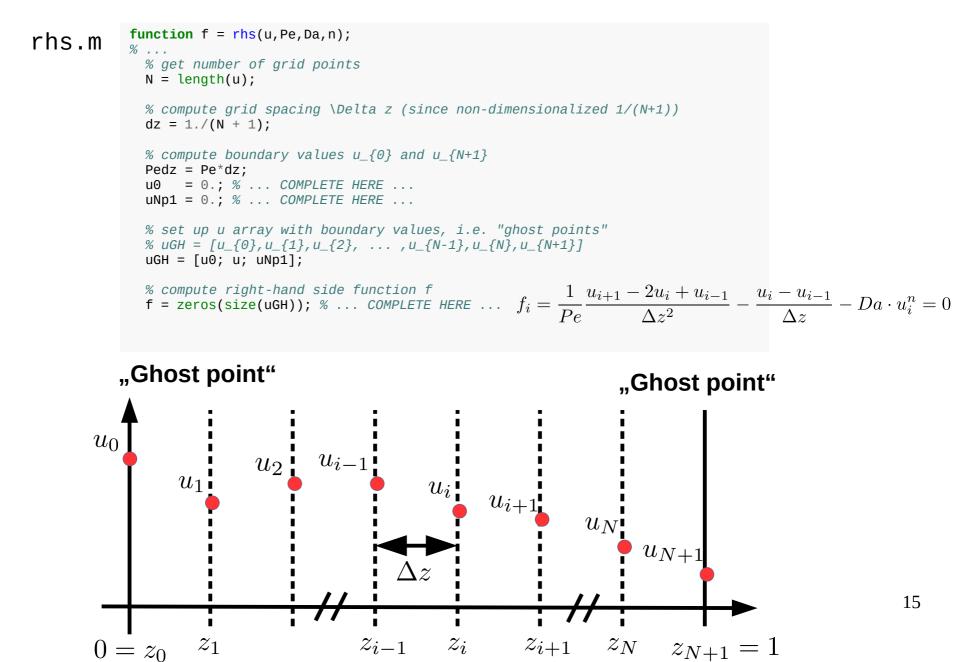
- Solve the steady state tubular reactor for 20 different Peclet numbers (between 0.01 and 100) and for a first (n=1) and a second (n=2) order reaction. Use a Damköhler number of unity. Complete the template rhs.m by implementing the non-linear equations to solve.
- 2. Plot the conversion at the end of the reactor $1 \frac{c_{\text{out}}}{c_{\text{in}}}$ vs. the Peclet number for both reaction orders. Also plot the ratio between the conversions of the first order and second order reaction

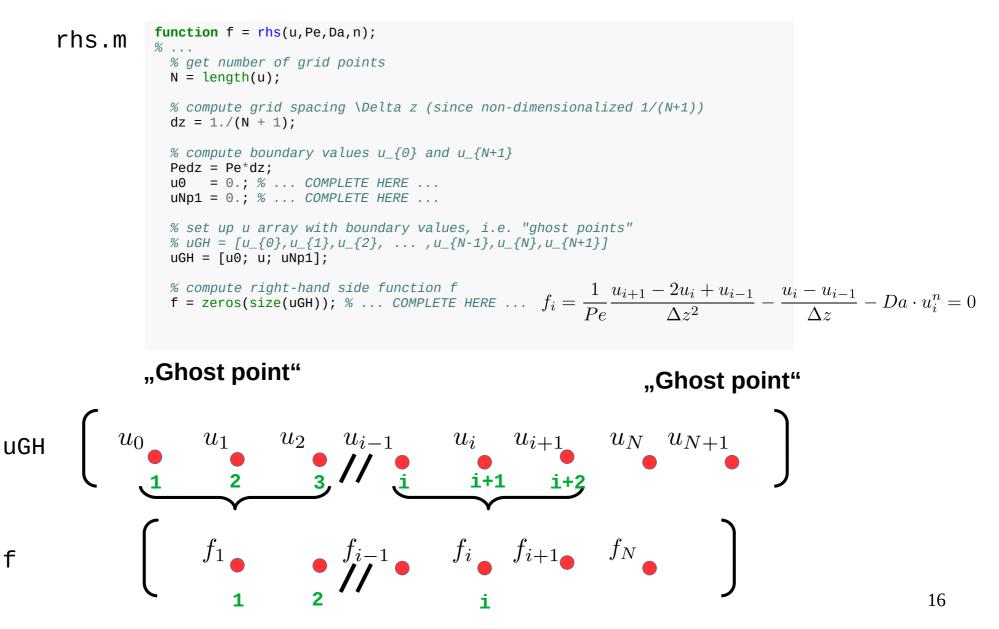
•What is better for these reactions, a lot of back-mixing (Pe small, CSTR) or ideal plug flow (Pe large, PFR)?

•What influence does the reaction order have overall and at low or high Peclet numbers?

Complete the template TubReact_steady_state.m

```
function f = rhs(u, Pe, Da, n);
rhs.m
            % function f = rhs(u);
            %
            % Purpose: compute the right-hand function of the spatially discretized
            %
                        (non-dimensionalized) advection-diffusion-reaction equation for a
            %
                        tubular reactor
            %
                         du/dtheta = - du/dz + 1/Pe d^2u/dz^2 - Da u^n
            %
            %
            %
                       using backward finite differences for the advection term and central
            %
                       finite differences for the diffusion term
            %
            % Input: u ... concentration
                     Pe ... Peclet number
            %
                     Da ... Damkoehler number
            %
                     n ... order of reaction
            %
            %
            % Output: f ... right-hand side function, i.e. du/dtheta
            %
            % Notes: None.
            %
              % get number of grid points
              N = length(u);
              % compute grid spacing \Delta z (since non-dimensionalized 1/(N+1))
              dz = 1./(N + 1);
              % compute boundary values u_{0} and u_{N+1}
              Pedz = Pe^*dz;
              u0 = 0.; \% \dots COMPLETE HERE.
                                                                                  Slide 12
              uNp1 = 0.; % ... COMPLETE HERE ...
              % set up u array with boundary values, i.e. "ghost points"
              % uGH = [u_{0}, u_{1}, u_{2}, ..., u_{N-1}, u_{N}, u_{N+1}]
              uGH = [u0; u; uNp1];
              % compute right-hand side function f
              f = zeros(size(uGH)); % ... COMPLETE HERE
```

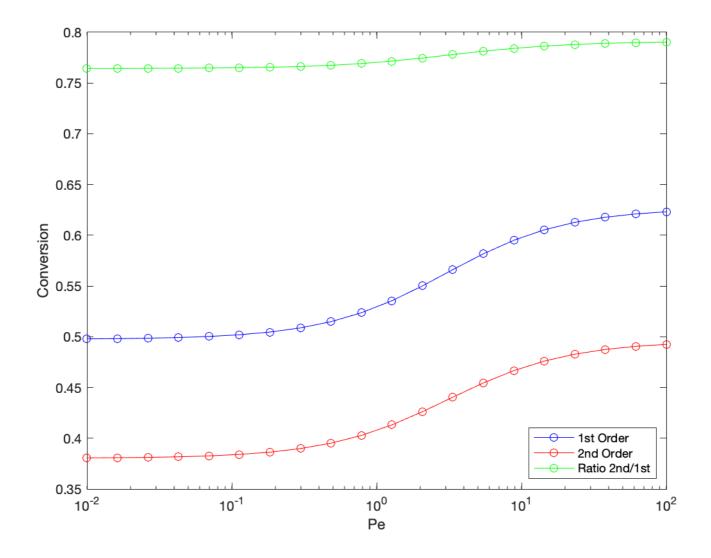


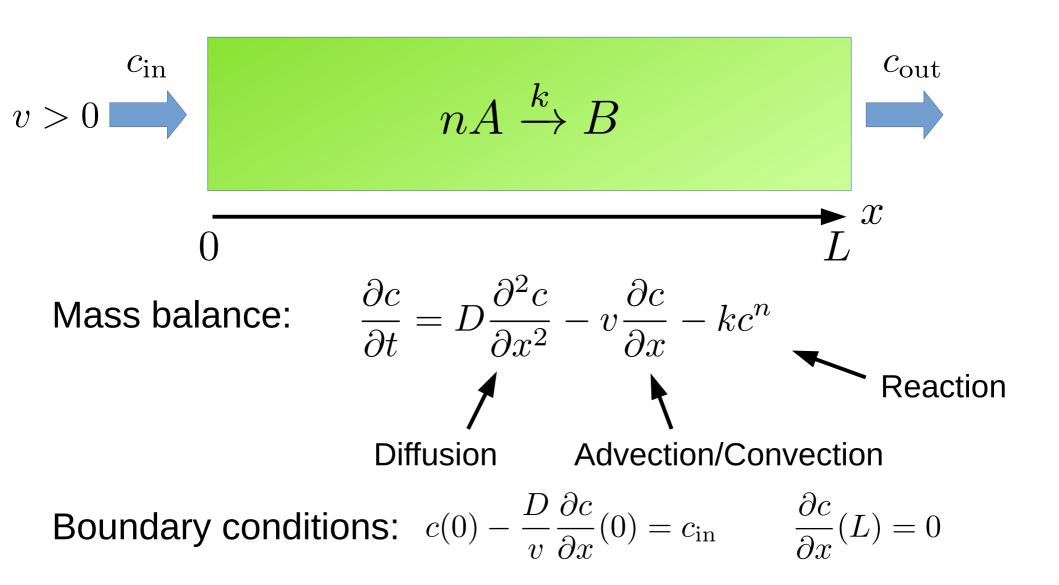


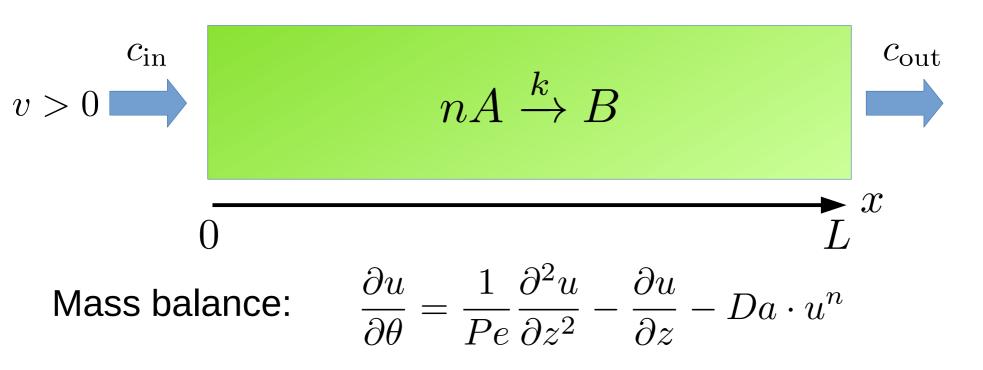
Array indexing (Matlab one-based indexing)

TubReact_steady_state.m

```
% set parameters
N = 100; \% number of grid points
NPe = 20; % number of Peclet numbers
Pe = logspace(-2,2,NPe); % generate NPe Peclet numbers log. spaced
                      % between 0.01 and 100
Da = 1.; % Damkoehler number
n = 1; % reaction order
% allocate array for concentrations
u = zeros(N, 1);
% allocate array for conversion & residuum
Conversion_n1 = zeros(NPe,1);
res_n1 = zeros(NPe, 1);
% solve BVP for all desired Peclet numbers
for iPe=1:NPe
 % ... COMPLETE HERE ...
                                                    Define a function handle and
 f = @(u) zeros(length(u), 1);
 res_n1(iPe) = norm(f(u), Inf);
                                                    use fsolve
end
% . . .
```



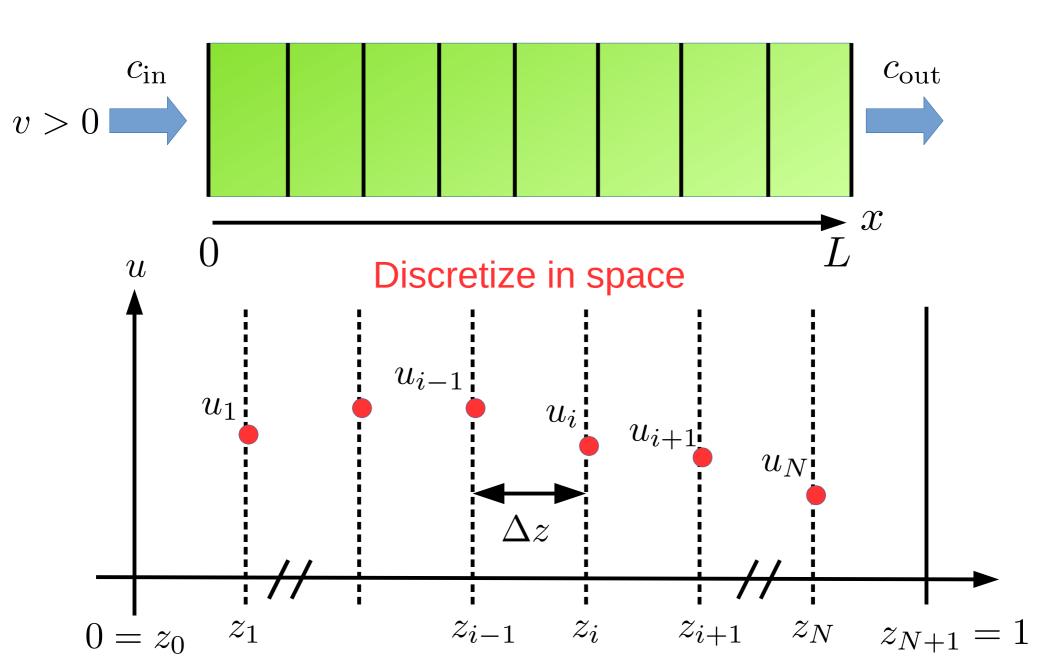


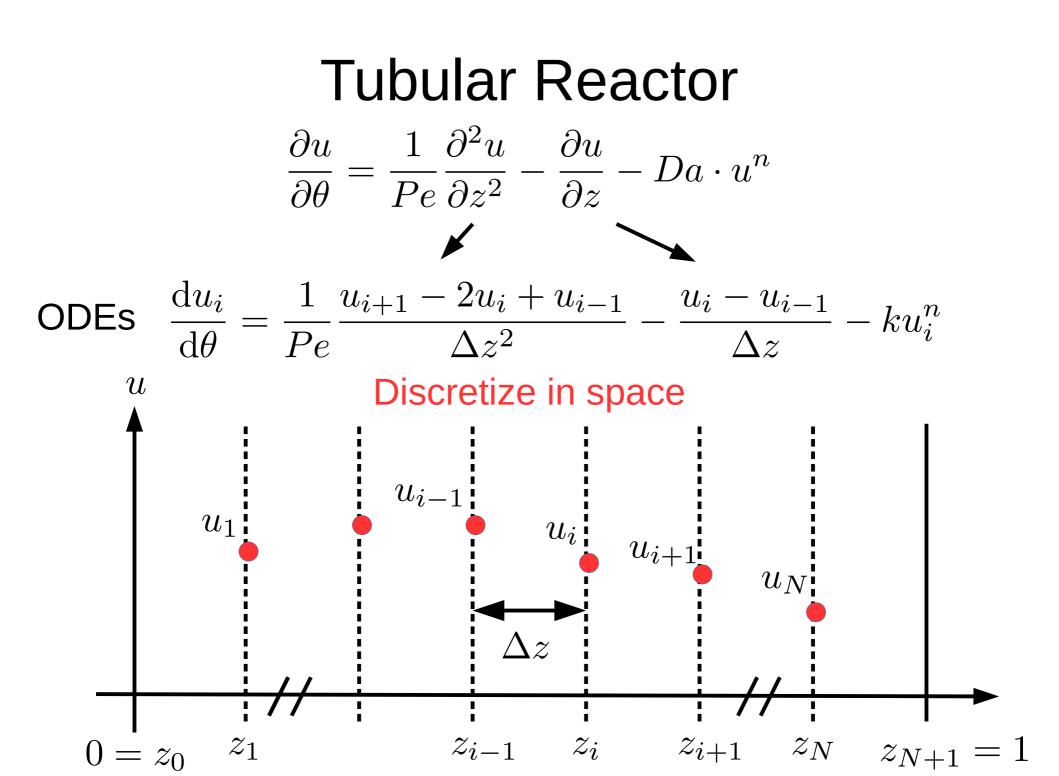


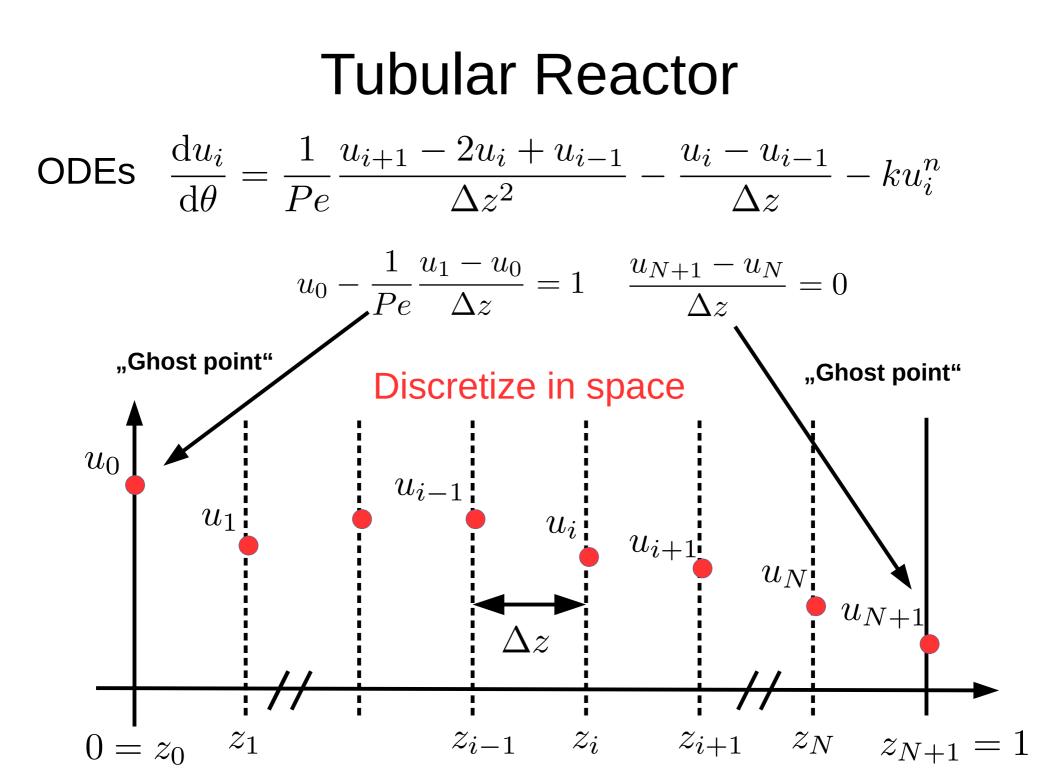
Dynamic tubular reactor

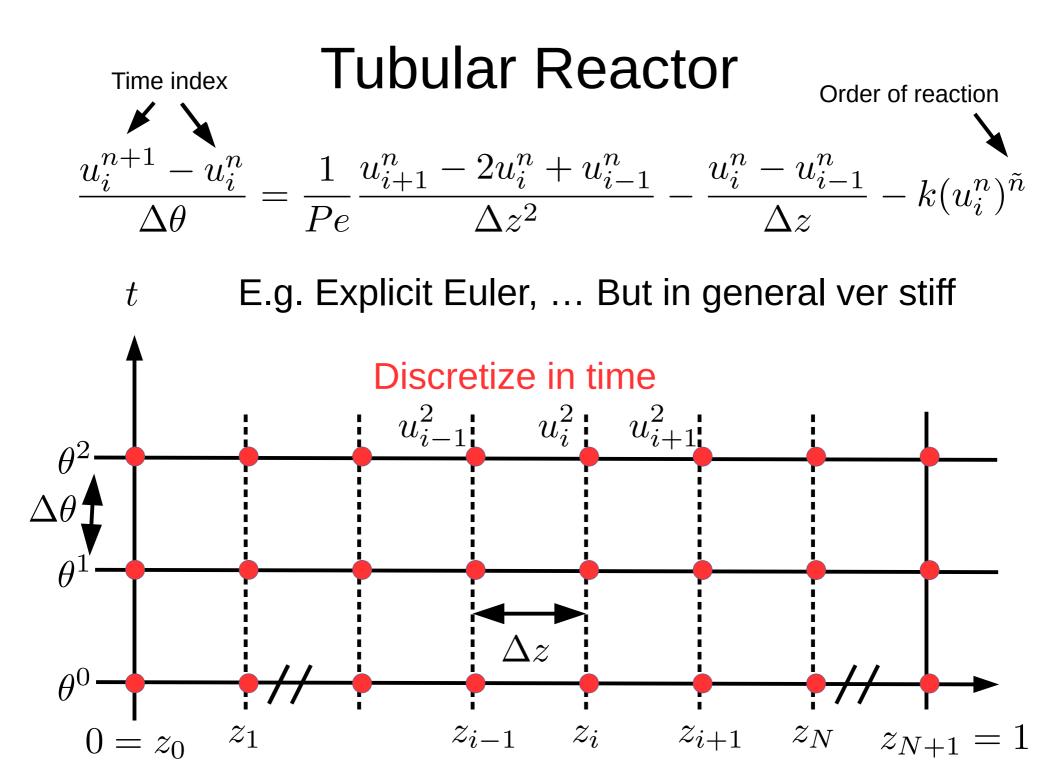
Boundary conditions:

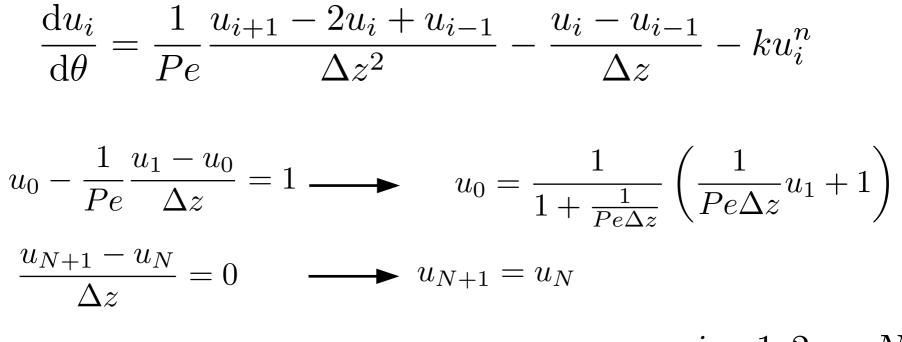
$$u(0) - \frac{1}{Pe} \frac{\partial u}{\partial z}(0) = 1$$
 $\frac{\partial u}{\partial z}(1) = 0$











i = 1, 2, ..., N

System of nonlinear ODEs!!! Stiff...

- 1.Solve the dynamic tubular reactor from initial 0 to final time of 5 with MATLAB's ode23s
 - Use the rhs.m from assignment 1 and the template TubReact_dynamic.m

Consider only a first order reaction with Pe=100 and Da=1

- 2.Plot the conversion at the end of the reactor vs. dimensionless time
- 3.At what time does the solution reach a steady state, i.e. how many reactor volumes of solvent will you need?

